

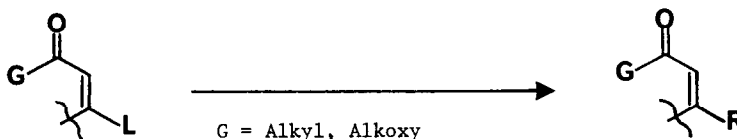
CHEMOSELECTIVE AND STERESELECTIVE ADDITION OF
 ORGANOCUPRATES TO α -OXOKETENE DITHIOACETALS

R. Karl Dieter*, Jeffrey R. Fishpugh, and Louis A. Silks

Department of Chemistry, Boston University, Boston, MA 02215

Abstract: The chemoselective and stereoselective addition of organocuprates to cyclic α -oxoketene dithioacetals affords predominately the E vinylogous thioesters. An acyclic α -oxoketene dithioacetal affords the Z vinylogous thioester stereoselectively. The observed selectivities are dependent upon substrate structure and the organocopper reagent.

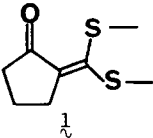
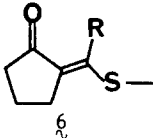
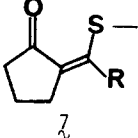
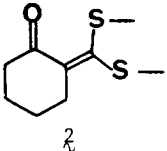
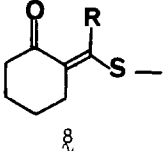
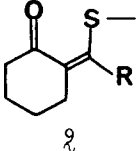
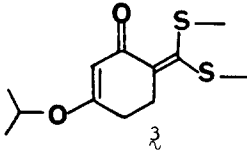
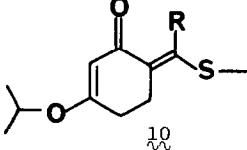
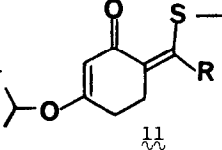
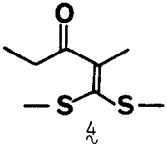
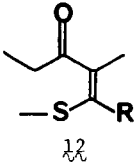
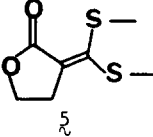
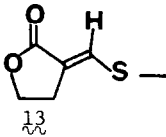
The chemoselective conjugate addition of organocuprates to α,β -unsaturated ketones and esters containing a good leaving group at the β -carbon atom constitutes a versatile synthetic route to β -substituted α,β -unsaturated ketones and esters. The reaction has been observed for



substrates containing halide,¹ acetate,² phosphate,³ alkoxy,^{4,5e} and alkylthio⁵ substituents as the leaving group. Generally, the substitution occurs stereoselectively in acyclic substrates^{1f,2,3,5b-d} and the procedure has been exploited for the stereoselective synthesis of trisubstituted olefins.^{2,3,5b-d} Similarly, α -oxoketene dithioacetals have been reported to undergo organocopper conjugate addition reactions in a relatively non-selective process to afford α -alkylidene or α -tertiary alkyl ketones.⁶ We report our observations detailing the chemoselective and stereoselective addition of organocuprates to α -oxoketene dithioacetals to afford vinylogous thioesters. This represents, to our knowledge, the first study of an efficient and selective addition of organocuprates to enones containing two good leaving groups at the β -carbon atom. The chemoselective conjugate addition of a single alkyl group permits the efficient utilization of the α -oxoketene dithioacetal functionality for sequential carbon-carbon bond formation.

Ketene dithioacetals **1** - **5** were readily prepared according to an established procedure.⁷ Initially, it was observed that some degree of chemoselectivity could be achieved by control of the reaction temperature. Treatment of enone **1** with 1.5 equivalents of lithium dimethylcuprate in ether at -50°C and -78°C afforded vinylogous thioester **6** in 46% and 65% yield respectively. It was anticipated that higher yields and greater chemoselectivity might be obtained by utilizing a more stable less reactive mixed cuprate. In the event, treatment of **1** with lithium phenylthiomethylcopper⁸ in THF⁹ (entry 1, Table 1) afforded **6** in 70% yield. Large scale experiments afforded the diastereomer **7** in less than 2% yield. Ketene dithioacetals **1** - **3** similarly underwent conjugate addition of lithium phenylthiomethyl (entries 4,5) or *n*-butylcopper (entry 2) to afford the E vinylogous thioesters as the predominate diastereomers.

Table 1. Conjugate Addition of Organocuprates to Conjugated Ketene Dithioacetals

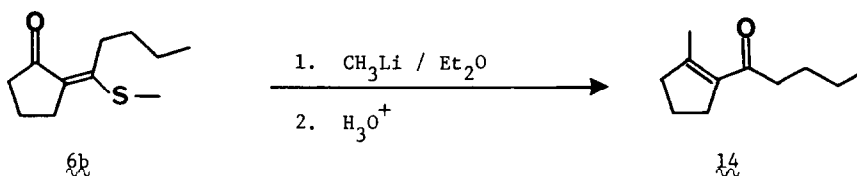
Entry	Substrate	R ^a	Product(s) ^b (% Yield) ^c	
				
1		a. CH ₃	70	2
2		b. <u>n</u> -Bu	65	—
3		c. (CH ₃) ₃ C	72 (1:1) ^d	
				
4		CH ₃	58	17
				
5		CH ₃	78	6
				
6		CH ₃	70	
				
7		CH ₃	60	

^aThe mixed cuprates [RCuSPh]Li were employed. Reactions were run in THF between -78°C and 0°C.^bSee reference 14. ^cAll yields are based upon isolated products purified by column chromatography on silica gel. ^dThis mixture of diastereomers was not separated.

Reaction of 1 with lithium phenylthio-*tert*-butylcopper afforded an equimolar mixture of diastereomers 6c:7c (entry 3). Entries 1-3 reveal that the stereoselectivity of the reaction is affected by the steric bulk of the organocopper alkyl group. Treatment of the acyclic substrate 4 with lithium phenylthiomethylcopper (entry 6) afforded a single vinylogous thioester which proved to be the Z isomer (12). Lactone 5 (entry 7) afforded the reduction product 13.

Diastereomers 6a, 7a and 8 - 11 were individually isolated and the geometry about the double bond initially determined from the proton magnetic resonance (pmr) spectra. The E vinylogous thioesters 6a, 8, and 10 displayed downfield vinyl methyl absorptions (δ 2.50, 2.38 and 2.45 respectively) compared to the Z vinylogous thioesters 7a, 9, and 11 (δ 2.15, 2.14 and 2.10 respectively). These chemical shift trends are in accord with the vinyl methyl absorptions reported for pulegone¹⁰ and α -isopropylidene-cyclopentanone.¹¹ The assignments were consistent with a study of the lanthanide induced shifts in the pmr spectra of 8 and 9. Additional evidence for the assignments was obtained from the UV spectra. The E diastereomers 6a, 8, and 10 gave more intense absorption bands at shorter wavelengths [$\lambda_{\text{max}}^{\text{ethanol}}$ (ϵ): 311 (24,000), 307 (18,000), and 321 (18,500) respectively] than the corresponding Z diastereomers 7a, 9, and 11 [$\lambda_{\text{max}}^{\text{ethanol}}$ (ϵ): 322 (16,000), 320 (12,000), and 332 (12,500) respectively] in accord with previous observations.¹² Structural assignments for vinylogous thioesters 6b and 7b were determined from the UV spectra [$\lambda_{\text{max}}^{\text{ethanol}}$ (ϵ): 312 (18,500) and 322 (13,500) respectively]. Vinylogous thioester 12 displayed a UV [$\lambda_{\text{max}}^{\text{ethanol}}$ (ϵ): 298 (12,000)] absorption suggestive of the Z isomer.

The pure vinylogous thioesters displayed a tendency to undergo isomerization upon standing as CCl_4 solutions or neat samples. Isomer 9 was observed to isomerize quantitatively to 8 upon standing in CCl_4 for several days, whereas 6b afforded a 96:4 mixture of 6b:7b. Similarly, the E/Z mixture of 6c:7c was slowly converted to the Z isomer [$\lambda_{\text{max}}^{\text{ethanol}}$ (ϵ): 318 (6,600)].



The chemoselective addition of organocuprates to α -oxoketene dithioacetals may be exploited as a key step in a 1,3-carbonyl transposition methodology that permits sequential regioselective carbon-carbon bond construction. Treatment of vinylogous thioester 6b with methyllithium followed by acid catalyzed hydrolysis¹³ affords transposed enone 14¹⁴ (60% yield). Consequently, a three step procedure is available for converting a ketone to a homologous β -substituted α,β -unsaturated ketone. The chemoselective and stereoselective addition of organocuprates to α -oxoketene dithioacetals, as described above, provides a versatile method for carbon-carbon bond constructions. Currently, investigations into other kinds of synthetic applications of this methodology are underway in our laboratory.¹⁵

Acknowledgements: We are pleased to acknowledge the partial support of this investigation by Research Corporation, the donors of the Petroleum Research Fund administered by the American Chemical Society and by Grant #IN-97E (Boston University School of Medicine) from the American Cancer Society.

REFERENCES AND NOTES

1. (a) E. Piers and H.E. Morton, J. Org. Chem., 44, 3437 (1979).
 (b) P.A. Wender and S.L. Eck, Tetrahedron Lett., 1245 (1977).
 (c) J.L. Coke, H.J. Williams, and S. Natarajan, J. Org. Chem., 42, 2380 (1977).
 (d) R.D. Clark and C.H. Heathcock, ibid., 41, 636 (1976).
 (e) F. Leyendecker, J. Drouin, and J.M. Conia, Tetrahedron Lett., 2931 (1974).
 (f) E. Piers and H.E. Morton, J. Chem. Soc., Chem. Commun., 1033 (1978).
2. (a) C. Ouannes and Y. Langlois, Tetrahedron Lett., 3461 (1975).
 (b) C.P. Casey and D.F. Marten, Synth. Commun., 3, 321 (1973).
 (c) C.P. Casey, D.F. Marten, and R.A. Boggs, Tetrahedron Lett., 2071 (1973).
3. (a) F.W. Sum and L. Weiler, Can. J. Chem., 57, 1431 (1979).
 (b) F.W. Sum and L. Weiler, Tetrahedron Lett., 707 (1979).
 (c) F.W. Sum and L. Weiler, J. Chem. Soc., Chem. Commun., 985 (1978).
4. S. Cacchi, A. Caputo, D. Misiti, Indian J. Chem., 12, 325 (1974).
5. (a) R.M. Coates and L.O. Sandefur, J. Org. Chem., 39, 275 (1974).
 (b) S. Kobayashi, O. Meguro-ku, and T. Mukaiyama, Chem. Lett., 705 (1974).
 (c) S. Kobayashi and T. Mukaiyama, ibid., 1425 (1974).
 (d) S. Kobayashi, H. Takei, T. Mukaiyama, and O. Meguro-ku, ibid., 1097 (1973).
 (e) G.H. Posner and D.J. Brunelle, J. Chem. Soc., Chem. Commun., 907 (1973).
6. (a) E.J. Corey and R.H.K. Chen, Tetrahedron Lett., 3817 (1973).
 (b) O.H. Johansen and K. Undheim, Acta Chem. Scand. (B), 33, 460 (1979).
7. R.K. Dieter, J. Org. Chem., 46, 5031 (1981).
8. G.H. Posner, C.E. Whitten, and J.H. Sterling, J. Am. Chem. Soc., 95, 7788 (1973).
9. The use of diethyl ether as solvent afforded lower yields of product and was generally unsatisfactory as a result of poor phenylthioalkylcuprate solubility.
10. K. Tori, I. Horibe, and H. Shigemoto, Tetrahedron Lett., 2199 (1975).
11. J.-M. Conia and P. Amice, Bull. Soc. Chim. Fr., 3327 (1968).
12. N. Furuhashi, M. Fukumura, T. Nishio, and S. Oae, Phosphorus and Sulfur, 5, 191 (1978);
 L.N. Owen and M.U.S. Sultanbawa, J. Chem. Soc., 3109 (1949).
13. S. Akiyama, S. Nakatsuji, T. Hamamura, M. Kataoka, and M. Nakagawa, Tetrahedron Lett., 2809 (1979).
14. All new products gave satisfactory ir, pmr, and cmr spectral data and C and H combustion analysis.
15. We are currently exploring the potential of this reaction as a method for effecting diastereoselective aldol condensations.

(Received in USA 19 May 1982)